

Ideal gas example in microcanonical ensemble -  
 Method of calculation of thermodynamic quantities  
 in microcanonical ensemble ~~take~~ by taking example  
 of classical ideal gas -

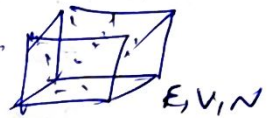
The Hamiltonian for ideal gas is given by

$$H(\vec{r}_i, \vec{p}_i) = \sum_{i=1}^N \frac{\vec{p}_i^2}{2m}$$

$$\Sigma(E) = \int d\vec{r}_i \int d\vec{p}_i$$

$$= V^N \int d\vec{p}_1 \int d\vec{p}_2 \dots \int d\vec{p}_N$$

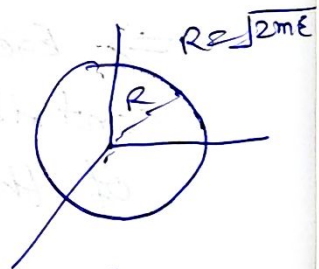
$$\frac{p_1^2}{2m} + \frac{p_2^2}{2m} + \dots + \frac{p_N^2}{2m} < E$$



~~System is fixed~~

or  $\Sigma(E) = V^N \Omega_{3N}(R)$

where  $R = \sqrt{2mE}$



3N-dim sphere

In the previous class we have calculated,

$$\Omega_n(R) = \frac{\pi^{n/2}}{(\frac{n}{2})!} R^n$$

Thus  $\Sigma(E) = V^N \frac{\pi^{\frac{3N}{2}}}{(\frac{3N}{2})!} (2mE)^{\frac{3N}{2}}$

Since we have

$$S(E, V, N) = \ln \Sigma(E)$$

we take  $k_B = 1$  for simplicity

$$S(E, V, N) = \ln V^N + \ln \pi^{\frac{3N}{2}} - \ln \left( \frac{3N}{2} \right)! + \ln (2\pi m E)^{\frac{3N}{2}}$$

or  $S = N \ln V + \frac{3N}{2} \ln \pi - \frac{3N}{2} \ln \frac{3N}{2} + \frac{3N}{2} + \frac{3N}{2} \ln (2\pi m E)$

we use Stirling approximation

$$\text{or } S = N \ln \left[ V \left( \frac{4\pi m E}{3N} \right)^{3/2} \right] + \frac{3}{2} N$$

$$\exp \left[ \frac{1}{N} \left( S - \frac{3}{2} N \right) \right] = V \left( \frac{4\pi m E}{3N} \right)^{3/2}$$

$$\text{or } E = \frac{3}{4\pi m} \frac{N}{V^{2/3}} \exp \left[ \frac{2}{3} \frac{S}{N} - 1 \right]$$

we have  $E(S, V, N)$ , using first law of thermodynamics (combined with second law) we can obtain the thermodynamic parameters  $T, P, \mu$  etc

$$T = \left. \frac{\partial E}{\partial S} \right|_{V, N} = \frac{1}{2\pi m} V^{-2/3} \exp \left[ \frac{2}{3} \frac{S}{N} - 1 \right]$$

$$P = - \left. \frac{\partial E}{\partial V} \right|_{T, N} = \frac{1}{2\pi m} \frac{N}{V^{5/3}} \exp \left[ \frac{2}{3} \frac{S}{N} - 1 \right]$$

or  $P V = N T$  Ideal gas equation of state